● PRINTER RUSH ● (PTO ASSISTANCE)

					HW
Application :	10/08370	Examiner :	Nutter	GAU:	1711
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	☐ OATH ☐ 312 ☒ SPEC	2-25-02			
The specification is binitled on 2-25-02 does not contain a paragraph References to color deamings or recurred per 37 CFR 1.84 (a)(2)(1). Please correct					
Thank You					
[XRUSH] RESPONSE: ATTACHEN IS A COPY OF THE BARACRAHP FUR COLOR PHOTO'S WHICH IS TO BE INSARTED IN TO THE SPECIFICATION UNDER THE BRIFE DESCRIPTION					
9-29-05					

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REV 10/04

INITIALS: \

[0021] Formula 6: $(M^1_4M^2_3A_3Y_2)_n$

[0022] Formula 6 represents a general formula for an embodiment of a polymer of the subject invention, as shown in Figures 11A-11F, wherein M^1 can be any metal that can sustain 3-fold rotational symmetry, wherein M^2 can be any metal that can sustain 4-fold rotational symmetry, wherein A is a trifunctional carboxylate with 3-fold rotational symmetry (allowing for geometric distortion), wherein Y is any -1 anion ("Y2" could also be just one "Y", if Y is a -2 anion), wherein 'n' indicates a polymeric structure in three dimensions (i.e., $n \ge 2$), and wherein any coordinating ligand or solvent molecule is optionally coordinated to each M.

[0023] Formula 7: (M₃A₂)_n

[0024] Formula 7 represents a general formula for another embodiment of a polymer of the subject invention, as shown in Figures 15A-15F, wherein M can be any metal that can sustain 4-fold rotational symmetry, wherein A is a trifunctional carboxylate with 3-fold rotational symmetry (allowing for geometric distortion), wherein 'n' indicates a polymeric structure in three dimensions (i.e., $n \ge 2$), and wherein any coordinating ligand or solvent molecule is optionally coordinated to each M.

[0025] In each of the above Formulas 1-7, M is a metal preferably in its 2+ transition state. However, it is also contemplated that M can be in other transition states (such as 1+, 3+, and so forth), and structures of the subject invention can contain M in more than one transition state (i.e., M(II)M(III)). For every M that is not in a 2+ transition state, there will preferably exist a counter ion to balance the charge (+ charge if < 2; - charge if > 2). The anions may, or may not, be coordinated to the metal.

SET > Brief Description of Drawings

[0026] Figures 1A-1C show cubohemioctahedron, small rhombihexahedron and small rhombidodecahedron uniform polyhedra, respectively, formed by linking vertices of squares only.

[0027] Figures 2A-2NN illustrate representative ligands for 120°.

[0028] Figures 3A-3G illustrate representative ligands for 144°.

[0029] Figure 4 shows the square nanoscale secondary building unit (nSBU), described by the general formula, $M_2(RCO_2)_4$, such as $[Cu_2(PhCOO)_4]$. Figure 4 (left) shows a ball-and-

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